# SYNTHESIS OF PLANT GROWTH REGULATOR ANALOGUES FROM ALEURITIC ACID

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Plant growth regulators (PGR'S) having either stimulating or retarding activity are non-pollutive and cost-effective compounds to bring about changes of desired productivity in agricultural crops. Among these compounds one group comprises of the aliphatic nature  $1^{\alpha-1}$ with remarkable and definite activity in the physiology of plant growth. This paper covers syntheses of two unreported PGR analogues (VIII) and (V) from threo-aleuritic acid<sup>2</sup>, 9, 10, 16-trihydroxyhexadecanoic acid (I), the major constituent acid of lac, believed to be present to the extent of ~35%.

(1), On periodate oxidation<sup>8</sup> gave quantitative yields of 7-hydroxy heptanal (III) and 9-oxononanoic acid (II). The latter on condensation with malonic acid in presence of pyridine on steam-bath for 8 hr gave 9-hydroxy -2- nonenoic acid (VI) which on esterification with MeOH-H<sup>+</sup> followed by mesylation yielded methyl 9-methylsulphonyloxy-2-nonenoate (VIII). Similarly II was converted into its methyl ester (IV) with diazomethane, which on condensation with malonic acid in the presence of pyridine gave 10-carboxy methyl-2-decenoic acid (V). The molecular ion peaks of V and VIII at 228 and 264 m/z confirmed the formation of the products. <sup>1</sup>HNMR spectrum of V and VIII showed doublets at  $\delta 5.83$  (J=16HZ) for ethylenic proton adjacent to carboxyl group and a multiplet at  $\delta$  5.54 for ethylenic proton attached to chain methylene groups.

$$\begin{array}{cccccc} H & OH \\ H_{11} & OH \\ H_{11} & CH_{2} & OH \\ (11) & CH_{2} & OH \\ (11) & CH_{2} & OH \\ H_{11} & OH \\ H$$

#### EXPERIMENTAL AND DISCUSSION

IR spectra were recorded on a Perkin-Elmer 237 spectrophotometer. Mass spectra on Jeol D-300 (Ei/ci) spectrometer with retention time 0.44 and 2.5. <sup>1</sup>HNMR spectra (chemical shifts in  $\delta$  ppm) on FTNMR instrument using TMS as internal standard. All the solvents were freshly distilled. Compounds were routinely checked by tlc.

9-Oxononanoic acid (II): (I) (30.4g, 100 m mol) was added to  $aq. Na_{2}CO_{8}$  solution (5.8%,110ml) with slight warming. After dissolution of (I), the solution was filtered and the filtrate

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was acidified with  $H_8PO_4$  (10%) till tutibidility appeared (pH ca 7). CHCl<sub>8</sub> (200 ml) was then added, the mixture was warmed to 30° when aq. NalO<sub>4</sub> (22g, 200 ml) was added over a period of 30 min. After stirring for 15 min., the mixture was cooled (5—10°), acidified to pH 3 with  $H_8PO_4$  (20% ~ 35 ml) and filtered. The organic layer was separated and extracted thoroughly with saturated NaHCO<sub>8</sub> solution (2 × 50 ml) followed by aq. Na<sub>2</sub>CO<sub>8</sub> solution (5%, 50 ml). The organic layer was then washed with water, brine and dried (Na<sub>2</sub>SO<sub>4</sub>). The combined aq. portions was acidified with  $H_8PO_4$  (10%) to pH 3 and extracted with CHCl<sub>8</sub>. Which on usual workup gave (II) (17g) as a thick liquid,  $\nu$  max 1728, 1700 cm<sup>-1</sup>  $j \in (CDCl_8)$ 1. 11-1. 82 (10 H, m, CH<sub>2</sub>COOH), 2.3 (4H, t, CH<sub>2</sub>CHO), 9.55 (1H, t, CHO), 9.87 (1H, bs, COOH).

7-Hydroxyheptanal (III): (III) (12g) was recovered from  $CHCl_8$  layer initially obtained under vacuum as liquid, 7 max 3250, 1720 cm<sup>-1</sup>;  $\delta(CDCl_8) 2.46(2H, m, CH_2CHO)$ 2.39 (1H, br, CH<sub>2</sub>OH, D<sub>8</sub>O exchangeable), 3.72 (2H, t, CH<sub>2</sub>OH), 9.91 (1H, t, CH<sub>2</sub>CHO).

Methyl 9-oxononanoate (IV): The ethereal solution of (II) was treated with fresh ethereal solution of diazomethane to obtain (IV) (yield 90%) also as an oil.  $\delta$ (CDCl<sub>8</sub>) 1.0-1.88 (10H, m, 5-CH<sub>2</sub>), 1.94-2.52(4H, m, CH<sub>2</sub>COOCH<sub>8</sub> and CH<sub>2</sub>CHO), 3.62(3H, s, COOCH<sub>8</sub>) 8.70 (1H, t, CHO).

#### 10-carboxymethyl 2-decenoic acid ( $\mathcal{V}$ ):

(IV) (5.8 g, 31.2 m mol) in dry pyridine (5g) was added dropwise to malonic acid (6.5 g, 62.5 m mol) in dry pyridine (8.7 g) and stirred for 10 min. The content was shaken for another 10-15 min and heated on steam-bath for 8 hrs till the evolution of CO<sub>2</sub> ceased. The reaction mixture was poured into water (25 ml) and the solution extracted with ether (2×200 ml). The ethereal solution was washed with dil HCl (25%), water and dried (Na<sub>2</sub>SO<sub>4</sub>). The ethereal solution was concentrated and eluted through a silicagel column with hexane— EtOAc (93:7). The pure (V) was recovered as viscus liquid (3.47g), 48.8%.  $\delta$  (CDCl<sub>8</sub>): 5.83 (IH, d, -CH=CHCO<sub>2</sub>H, J=16HZ), 5.54 (1H, m, CH-CHCO<sub>2</sub>H, 3.66 (3H, s-CH<sub>8</sub>OCO), 2.32 (2H, t, CH<sub>2</sub> OCOCH<sub>8</sub>; J=8HZ), 2.23(2H, q-CH<sub>2</sub>-CH=CH), 1.33-1.62 (10H, br, (CH<sub>2</sub>)n), MS (m/z) 228(M<sup>+</sup>0.7).

Methyl 9-hydroxy-2-nonenoate (VII): 9-Hydroxy-2-nonenoic acid (VI) (4.36 g, 71.5%) was obtained from III (4.6 g, 35.4 m mol) by adopting the same procedure as for (IV). (VI) (3.1g, 18m mol) was esterified with dry MeOH. BF<sub>8</sub>. Et<sub>2</sub>O complex (51 ml) by refluxing for 15 min. Usual work-up afforded the title compound (VII) (.34g, 40%) as a liquid showing single spot on tlc (Hexane, EtOAc-1:1).

Methyl 9-methylsulphonyloxy-2-nonenoate (VIII) (1.2g, 6.5 m mol) NEt<sub>8</sub> (0.98g, 9.8 m mol) and CH<sub>2</sub>Cl<sub>2</sub> (15 ml) were well mixed and cooled to O°. Mesylchloride (0.8186g, 7.15 m mol) was added drop wise to the cooled reaction mixture during 30 min. The reaction mixture was transferred to a separating funnel with additional 10 ml of CH<sub>2</sub>Cl<sub>2</sub> and washed with ice-water, cold HCl (10%), saturated NaHCO<sub>3</sub> soln. and brine sequentially,

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dried (Na<sub>2</sub>SO<sub>4</sub>). Removal of the solvent furnished VIII as a liquid (1.42g, 83%) showing single spot on the (Hexane, EtOAc-10:3).

 $\delta(\text{CDCl}_3)$ : 5.84 (IH, d,  $-\text{CH} = \text{CH} - \text{CO}_2$ H, J = 16 HZ), 5.54(IH, br, -CH = CH - CO<sub>2</sub>H),4.23 (2H, t,  $-\text{CH}_2 - \text{OSO}_3$  CH<sub>3</sub>, J = 6HZ), 3.74 (3H, s,  $-\text{COOCH}_3$ ), 3.02(3H, s,  $-\text{CH}_3$  SO<sub>2</sub>O), 2.34 (2H, t,  $-\text{CH}_2 - \text{OSO}_2$ CH<sub>3</sub>), 2.22 (2H, q,  $-\text{CH}_2 - \text{CH} = \text{CH} -$ ), 1.38 - 1.8(10H, m, - (CH<sub>2</sub>)n); MS (m/z) : 264(M<sup>+</sup>, O.9), 233(M - OCH<sub>3</sub>, 9.9), 2O5(M - CO<sub>2</sub>CH<sub>3</sub>, O.4), 185 (M - SO<sub>2</sub>CH<sub>3</sub>, 2.6).

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#### SUMMARY

Plant glowth regulator analogues, methyl 9-methylsulphonyloxy-2 nonenoate (VIII) and 10-carboxymethyl 2-decenoic acid (V) were synthesised from 7-hydroxyheptanal (III) and 9-oxononanoic acid (II), the periodate oxidation products of *threo*-aleuritic acid (I), the major constituent acid of lac ( $\sim 35\%$ ) adopting simple reaction sequences. The products were <sup>1</sup>HNMR and mass spectra,

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